Palladium-Catalyzed Allylation of Pyrimidine-2,4-diones (Uracils) and of 6-Membered Heterocyclic Ambident Sulfur Nucleophiles

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Abstract.- Pd(0)-Catalyzed allylation of six-membered ambident heterocycles bearing NH-CO, NH-CS and CH₂-CO moleties obey the regioselectivity rules C>O, N>O, S>N, NH-CO>NH-CS Uracil and 5-methyluracil (thymine) do not show regioselectivity (N-1 = N-3) whereas 6-methyluracil is regioselectively allylated at N-3 (N-3>N-1)

INTRODUCTION

The Pd(0)-catalyzed allylation of heterocyclic systems bearing ambident nucleophiles is a topic of increasing interest. Thus, regioselective N-9 allylation of purines, at the imidazole part of the molecule, is a key step in the preparation of carbanucleosides 1-7 Ambident heterocyclic 6-membered rings possessing a tautometric or mesometric aromatic structure that have been allylated under Pd(0) catalysis include 2-pyridones (N-allylation under Pt(0) catalysis),⁸ pyrimidin-2-ones (N-allylation under thermodynamic control), 1,9,10 pyridine-2,6-diones (C-allylation), 11 pyrimidine-2,4-dione (uracil) (N-1 allylation), 12,13 and its 5-methylderivative (thymine) (N-1 allylation), 13 4-hydroxy-6-methyl-2-pyrone (triacetic acid lactone) (C-allylation under thermodynamic control), 14,15 and barbituric acid 16 However, a very recent paper⁷ reports the regioselective Pd(0)-catalyzed allylation of citosine and 5-methylicitosine to occur at N-1 but the allylation of thymine with cyclopentadiene monoepoxide to be non selective, both products from allylation at N-1 and N-3 being produced

On the other hand, sulfur nucleophiles are not popular in Pd(0)-catalyzed allylation chemistry, possibly due to the belief that the pronounced thiophilicity of palladium could poison the catalytic systems. However, Trost and Scanlan have described a Pd(0)-catalyzed synthesis of allyl sulfides¹⁷ and also two scattered examples of allylations at sulfur under Pd¹⁸ and Ni¹⁹ catalysis have been reported.

RESULTS

We have studied the Pd-catalyzed allylation of several 6-membered heterocyclic ambident systems bearing a nucleophilic sulfur atom, 1 e 2-thiopyridone, 3, 2-mercaptopyrimidine, 4, 6-methyl-2-thiouracil, 6, and 2thiobarbituric acid, 7 Some oxo analogues such as 2-pyridone, 2, 6-methyluracil, 5, thymine, 8, and uracil, 9 were also included Cinnamyl ethyl carbonate, 1a, was chosen as allylic reagent for its high regioselectivity ¹⁶

Our results are collected in the table and the scheme Pyridone, 2, affords the N-cinnamyl derivatives 10 (runs 1 and 2), thus pointing out to a preferential attack at the nitrogen atom of the amide (N>O) as previously observed under Pt catalysis ⁸ However, the thioanalogue 3 behaves differently to give sulfides 11 (runs 3 and 4) Thus, the conjugate base of the thioamide (or its tautomer) exhibits preferential attack by the sulfur nucleophilic center (S>N) Clearly the sulfur atom is compatible with the initial presence of a Pd(II) species Also, 2-mercaptopyrimidine, 4, was allylated at the sulfur atom giving 12 (run 5)

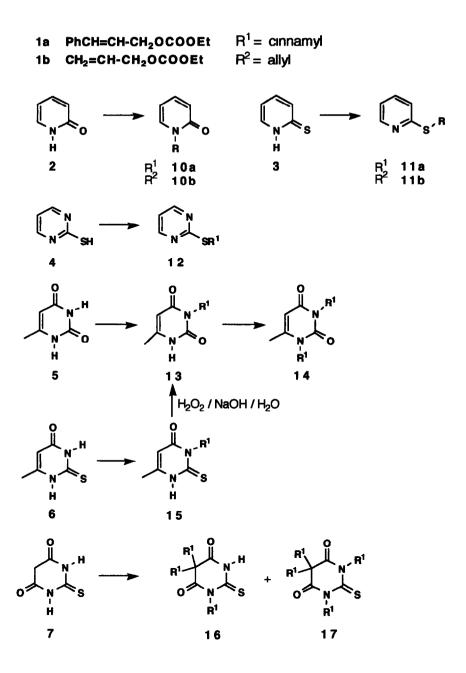
Next, we studied compounds 5 and 6. 6-Methyluracil, 5, affords N-cinnamyl derivative 13 (N>O, N-3 > N-1) (run 6) Further reaction of 13 produced the dicinnamyl derivative 14 (run 8) However, the sulfur counterpart 6 produces 15 as the only clearly identified isomer (run 9) Compound 15 was converted into 13 by oxidation with hydrogen peroxide, thus giving an additional evidence that both compounds have the cinnamyl group at the same position.

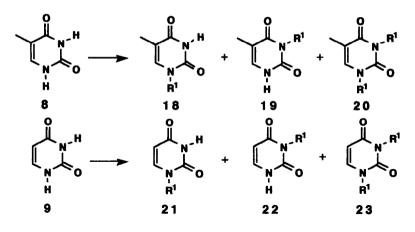
2-Thiobarbitum acid, 7, bears a nucleophilic carbon atom and, as already observed for barbitum acid, ¹⁶ the active methylene group 1s the preferred point for reaction followed by the nitrogen atoms (C>O, N>O, CH₂-CO > NH-CO, NH-CO > NH-CS) as pointed out by the isolation of **16** and **17** (run 10)

Run	2-9,13 (mmole)	1 (mmole)	Pd ^a (mmole)	Solvent (mL)	temp	time	Products (%)	Allylation
1	2 (10 0)	1a (12 0)	A(0.5)	THF(20)	rt	16h	10a (88)	N > 0
2	2 (7 0)	1b(8.4)	A(0.35)	THF(15)	rt	16h	1 0b (70)	N > O
3	3(70)	1a(84)	A(035)	THF(20)	rt	64h	11a(96)	S > N
4	3(10 0)	1b (10 0)	B(0 5)	THF(20)	rt	64h	11b(80)	S > N
5	4(8.0)	1a(8.0)	B(0.4)	DMSO(20)	rt	2 5h	12(55)	S > N
6	5 (8 0)	1a (96)	A(0 4)	Diox (20)	Reflux	24h	13(52)	N > O, N-3 > N-1
7	5 (8.0)	1a (80)	B(0.4)	DMSO(25)	1 05 C	14h	1 3 (49), 14(5)	N > O, N-3 > N-1
8	13 (0 8)	1a (1 0)	A (0 04)	THF (20)	Reflux	3h	14(90)	N > 0
9	6 (8 0)	1a (9.6)	A(0.4)	Diox.(20)	Reflux	6d	15(30)	N > O NHCO > NHCS
10	7(4.0)	1a (13 0)	B(0 4)	THF(20)	rt	16h	16(20), 17(32)	C > N, N > O, NHCO > NHCS
11	8 (4 0)	1a (4 0)	B(0 2)	DMSO(20)	1 08 C	45h	18 (30), 19 (14),	N-1 = N-3
12	9 (9 0)	1a (90)	B(0 4)	DMSO(20)	1 05 C	5 5h	20 (7) 21 (38),	N-1 = N-3
		()					22 (7), 23 (9)	

Table - Allylation of Compounds 2-9 and 13

a A Pd(acac)2/PPh3 (1 4), B Pd(PPh3)4





Scheme For experimental conditions see table

The reaction of 5 at N-3 (run 6) was surprising in view of the literature precedents dealing with uracil and thymine as indicated above 12,13 Therefore, we decided to perform some experiments on thymine, 8, and uracil, 9 Indeed, reactions of both 8 and 9 with 1a (runs 11 and 12) showed poor regioselectivity, reactions at N-1 being only slightly predominant and allylation at N-3 being significant A very recent paper by a Hoechst AG group reports similar results.⁷ Our reactions with 8 and 9 were performed in DMSO for solubility reasons and therefore we performed a second experiment with 5 in the same solvent (run 7) Again, the result was a remarkable regioselectivity favouring N-3 Possibly the steric hindrance introduced by the methyl group at C-6 is responsible for the observed results. In any case the conclusion is that extreme caution in structure determinations is required in this field

All N-cinnamyl derivatives present a doublet at d 4 4-5 2, whereas S-cinnamyl derivatives exhibit the doublet at 3 9-4 1. Furthermore, the signals for the S-<u>CH</u>2 appear in 13C-NMR at 32 7-33.0 whereas those for N-<u>CH</u>2 appear above 40 Assignments of structure to monoallylated products were made on the basis of the 13C-NMR spectra with the program Selective Distorsionless Enhancement by Polarization Transfer (SDEPT) developed by Sánchez-Ferrando and coworkers in our Department ²⁰ As an example, by selective pulsing of the methylene protons of compound **13** only the coupled carbonyl carbon atoms (apart from the cinnamyl olefinic carbon atoms) at C-2 and C-4 showed signals enhanced by polarization transfer (SDEPT effect) Should the isomeric structure (cinnamyl group at N-1) have been the real one, SDEPT effect for signals due to C-2 and C-6 would have been observed A similar technique (Selective INEPT) has been recently reported ²¹

FINAL REMARCK

When this work was already finished Prof Denis Sinou (Lyon) announced us that he and his coworkers had obtained results similar to those here described We are indebted to him for this communication ²⁷

EXPERIMENTAL

<u>N-Cinnamyl-2-pyrdone</u>, **10a** (Run 1) (General procedure) A degassed solution of Pd(acac)₂ (0 152 g, 0.5 mmole), triphenylphosphine (0.525 g, 20 mmole) and cinnamyl ethyl carbonate (2 475 g, 120 mmole) in anhydrous THF (15 mL) was added over a degassed solution of 2-pyridone (0.951 g, 10.0 mmole) in anhydrous THF (5 mL) The stirred mixture was kept 16 h under argon at room temperature The formed yellow solid (a palladium complex) was filtered off The filtrate was evaporated and the residue was chromatographed through a column of silica-gel to afford **10a** (1 863 g, 88%), b p 225C/4 5 mmHg, IR(film). 1658 cm⁻¹, 1H-NMR (CDCl₃) 4 75 (d, J = 5 7 Hz, 2H), 6 22 (dt, J = 8 2 and 1 9 Hz, 1H), 6.31 (dt, J =

150 and 57 Hz, 1H), 6.59 (dd, J = 82 and 1.9 Hz, 1H), 666 (d, J = 15.0 Hz, 1H), 722-787 (m, 7H), 13C-NMR (CDCl₃) 504, 105.9, 1207, 1230, 126.3, 1278, 1283, 1338, 1358, 1368, 1392, 1622, MS (m/e) 212(M+1, 7), 211(M, 43), 120(33), 117(85), 116(53), 115(100), 96(41), 91(32). <u>Anal</u> Calcd for C14H₁₃NO C, 7959, H, 620; N, 663 Found C, 7967; H, 626; N, 639

All other compounds were prepared as for **10a** under the particular conditions described in the table <u>N-Allyl-2-pyridone</u>, **10b** (Run 2) The final oil was not distilled but converted into the picrate m p 105-6C (Lit 22 m p 104 5-105 5C)

Free 10b IR(film). 1657 cm^{-1} , 1H-NMR (CDCl₃) 451 (d, J = 48 Hz, 2H), 490-520 (m, 2H), 547-600 (m, 1H), 612 (dt, J = 74 and 18 Hz, 1H), 651 (d, J = 78 Hz, 1H), 711-742 (m, 2H), 13C-NMR (CDCl₃) 500, 1052, 1173, 1198, 1318, 1367, 1387, 1614, MS(m/e) 135(M, 69), 134(100), 120(82), 79(37), 41(41)

<u>2-(C1nnamylthio)pyridine</u>, <u>11a</u> (Run 3) The filtrated Pd complex was bis(pyridine-2-thiolate)bis(triphenylphosphine)palladium, mp 181-2C (Lit ²³ mp 185-6C) The filtrate was evaporated and the residue was distilled to give 1 533 g (96%) of **11a**, b p 150C/0 05 mmHg, mp 41-2C (Lit ²⁴ mp 42-42 5C), 1H-NMR (CDCl₃) 408 (d, J = 6 1 Hz, 2H), 6 32 (dt, J = 16 0 and 6 1 Hz, 1H), 6 68 (d, J = 16 0 Hz, 1H), 6 68-7 68 (m, 8H), 8 48 (d, J = 5 0 Hz, 1H), 13C-NMR (CDCl₃) 32 7, 119 4, 122 4, 125 4, 126 3, 127 4, 128 4, 132 7, 135.9, 136 9, 149 3, 158 6, MS (m/e) 227(M, 61), 194(95), 136(29), 117(89), 115(100), 91(33)

<u>2-(Allylthio)pyridine, 11b (Run 4)</u> Isolated as an oil. Its spectroscopic data were in agreement with those already described $2^{5,26}$ 1H-NMR (CDCl₃) 3 82 (d, J = 6 9 Hz, 2H), 5 09 (d, J = 9 8 Hz, 1H), 5 27 (dd, J = 17 2 and 2 2 Hz, 1H), 5 95 (ddt, J = 17 2, 9 8 and 6 9 Hz, 1H), 6 94-6 99 (m, 1H), 7 16 (d, J = 8 0 Hz, 1H), 7 47 (dt, J = 2 0 and 8.0 Hz, 1H), 8 38 (d, J = 6 4 Hz, 1H), 13C-NMR (CDCl₃) 32 8, 117 3, 119 3, 122 1, 128 1, 135 7, 149 2, 158 4, MS(m/e) 151(M, 25), 136(100), 118(32), 79(51)

<u>2-(Cinnamylthio)pyrimidine, 12 (Run 5)</u> The filtrated palladium complex was bis(pyrimidine-2-thiolate)bis(triphenylphosphine)palladium (13%), mp 177-8C, 1H-NMR (d6-DMSO) 691 (d, J = 49 Hz, 2H), 73-75 (m, 30H), 821 (d, J = 49 Hz, 4H) <u>Anal</u> Calcd for C44H36N4P2PdS2 = (C4H3N2S)2Pd(PPh3)2) C, 61 94, H, 425, N, 657, S, 752 Found C, 62 02, H, 422, N, 653, S, 730 The filtrate was evaporated and the residue was distilled to afford 12 p 100-125C/0 2 mmHg, m p 32-3C, 1H-NMR (CDCl3) 3 95 (d, J = 62 Hz, 2H), 631 (dt, J = 155 and 62 Hz, 1H), 666 (d, J = 155 Hz, 1H), 692 (t, J = 45 Hz, 1H), 730 (m, 5H), 8 50 (d, J = 45 Hz, 2H), 13C-NMR (CDCl3) 330, 1162, 1245, 126 0, 127 2, 128 2, 132 6, 136 4 156 8, 171 6, MS(m/e) 228(M, 16), 195(100), 117(50), 115(52) <u>Anal</u> Calcd for C13H12N2S C, 68 39, H, 530, N, 12 27, S, 14 04 Found C, 67 80, H, 524, N, 11 82, S, 13 28

<u>3-Cinnamyl-6-methyluracyl</u>, <u>13</u> (Run 6) No Pd complex was isolated in this reaction Product 13 precipitated when the reaction mixture was reaching room temperature. It was filtrated off and the filtrate was evaporated to afford a mixture of 13 and the starting material 5 Compound 13 has m p 233-4C (from ethanol), IR(KBr) 1735, 1644, 1609 cm⁻¹, 1H-NMR (d6-DMSO) 2 03 (s, 3H), 4 49 (d, J = 6 0 Hz, 2H), 5 48 (s, 1H), 6 23 (dt, J = 15 5 and 6 0 Hz, 1H), 6 46 (d, J = 15 5 Hz, 1H), 7 22-7 38 (m, 5H), 13C-NMR (d6-DMSO) 179, 407, 98.2, 124 2, 126 0, 127 4, 128 4, 131 5, 136 2, 151 1, 151 2, 162 4, MS(m/e) 242(M, 25), 151(100), 127(28), 115(29) Anal Calcd for C14H14N2O2 C, 69 40, H, 5 82, N, 11 56 Found C, 69 50, H, 5 73, N, 11 53

<u>1,3-Dicinnamyl-6-methyluracil, 14 (Run 8)</u> No Pd complex was isolated in this reaction. Compound 14 was isolated by solvent evaporation and column chromatography through silica-gel. Compound 14 has m p. 125-6C (ethanol), IR(KBr) 1693, 1651, 1623 cm⁻¹, 1H-NMR (CDCl₃) 2.35 (s, 3H), 4.74 (d, J = 5 Hz, 2H), 4.84 (d, J = 5 Hz, 2H), 5.72 (s, 1H), 6.06-6.50 (m, 2H), 6.61 (d, J = 15 Hz, 1H), 6.79 (J = 15 Hz, 1H), 7.35 (m, 10H), 13C-NMR (CDCl₃) 19.5, 42.9, 46.6, 101.9, 123.1, 126.5, 127.6, 128.1, 128.4, 128.6, 133.0, 134.0, 135.9, 136.7, 151.4, 151.9, 161.8, MS(m/e) 3.58(M, 20), 241(20), 132(23), 117(40), 115(20), 115

115(100), 110(32), 91(19) <u>Anal</u> Calcd for C₂₃H₂₂N₂O₂ C, 77 07, H, 6 19, N, 7 81 Found C, 77 10, H, 6 17, N, 7 81

<u>3-Cinnamyl-6-methyl-2-thiouracil, 15 (Run 9)</u> No Pd complex was isolated Product 15 was separated by digestion with diethyl ether. The residue crystallized from ethanol to afford 15, mp 201-2C, IR(KBr) 1668, 1664, 1623 cm⁻¹, 1H-NMR (d6-DMSO) 2 13 (s, 3H), 5 03 (d, J = 5 3 Hz, 2H), 5 83 (s, 1H), 6 23 (dt, J = 16 7 and 5 3 Hz, 1H), 6 57 (d, J = 16 7 Hz, 1H), 7 13-7 50 (m, 5H), 13C-NMR (d6-DMSO) 17 8, 46 6, 102 7, 122 9, 126 0, 127.4, 128 4, 132 4, 136 1, 151 6, 159 8, 176 3, MS(m/e) 258(M, 36), 225(100), 149(21), 117(66), 115(95) Anal Calcd for C14H14N2OS C, 65 09, H, 5 46, N, 10 84, S, 12 41 Found C, 65 02, H, 5 47; N 10 82, S, 12 33

<u>Preparation of 13 by oxidation of 15</u> 36% Hydrogen peroxide (0.300 g, 3 30 mmole) was dropwise added under stirring at room temperature upon a mixture of 15 (0.200 g, 0 08 mmole) and 20% aqueous sodium hydroxide (20 mL) After 3 h some starting material still remained and more 36% hydrogen peroxide (2 mL) was added The mixture was stirred for 24 h and partitioned with dichloromethane The organic layer was washed with aqueous sodium bisulfite, with aqueous HCl and with water, then it was dried and evaporated to afford a residue that upon digestion with diethyl ether gave 13 (0 105 g, 53%), m p 233-4 C

<u>1.5.5-Tricinnamyl-2-thiobarbituric acid</u>, <u>16</u>, and <u>1.3.5.5-tetracinnamyl-2-thiobarbituric acid</u>, <u>17</u> (Run 10) After solvent evaporation the residue was digested with ether to afford an insoluble Pd complex of m p 135-6 C (from acetone) The ether solution was evaporated to give an oil that was chromatographed through a column of silica-gel The following products were eluted by increasing the solvent polarity

17, m p 61-2C (diethyl ether), IR(KBr) 1693 cm⁻¹, 1H-NMR (CDCl₃) 2 97 (d, J = 7 7 Hz, 4H), 5 17 (d, J = 6 3 Hz, 4H), 5 8-6 9 (m, 8H), 7 16 and 7 24 (two s, 20H), 13C-NMR (CDCl₃) 42 5, 49 3, 58.1, 121 4, 121 9, 126 2, 126 4, 127 6, 128 3, 134 9, 135 6, 136 2, 168 9, 178 8, MS(m/e) 91(100) <u>Anal</u> Calcd for C40H₃₆N₂O₂S C, 78 92, H, 5 96, N, 4 60, S, 5 27 Found C, 78 60, H, 6 03, N, 4 61, S, 4.80

16, m p 48-9C, IR(KBr) 1725, 1689 cm⁻¹; 1H-NMR (CDCl₃) 2 97 (d, J = 7 7 Hz, 4H), 5 05 (d, J = 6.3 Hz, 2H), 5 8-6 9 (m, 6H), 7 21 (s, 15 H), 9 05 (s, 1H), 13C-NMR (CDCl₃) 41.8, 47 9, 58 2, 121 1, 121 2, 126 2, 126 4, 127 6, 128 3, 135 3, 135 8, 136 0, 136 2, 168 1, 169 5, 177 3, MS(m/e). 227(41), 212(24), 194(67), 149(23), 136(25), 117(91), 115(100), 91(60) <u>Anal</u> Calcd for C₃₁H₂₈N₂O₂S C, 75 58, H, 5 73, N, 5.69, S, 6 51 Found C, 75 57, H, 5 76, N, 5 64, S, 6 40

<u>1-Cinnamyl-5-methyluracil, 18, 3-cinnamyl-5-methyluracil, 19, and 1.3-dicinnamyl-5-methyluracil, 20 (Run 11)</u> After solvent evaporation at water pump pressure, the residue was partitioned between dichloromethane and water The organic layer was dried and evaporated The residue was digested with diethyl ether to afford 18, m p 210-1C (from ethanol), IR(KBr) 1695, 1676 cm⁻¹, 1H-NMR (d6-DMSO) 1 72 (s, 3H), 4 39 (d, J = 6 1 Hz, 2H), 6.31 (dt, J = 15 9 and 6 1 Hz, 1H), 6 53 (d, J = 15 9 Hz, 1H), 7 24-7 42 (m, 5H), 11 25 (s, 1H), 13C-NMR (d6-DMSO) 12 5, 48 5, 108 9, 124 6, 126 7, 127 9, 128 7, 132 5, 135 8, 140 9, 150 8, 164 3, MS(m/e) 242(M, 23), 117(100), 115(36) Anal Calcd for C14H14N2O2 C, 69.40, H, 5 82, N, 11 56 Found C, 69 39, H, 5 90, N, 11 42

The ether solution was evaporated and the residue chromatographed through a column of silica-gel to afford

20 m p 162-3C (from diethylether-dichloromethane), IR(KBr) 1694, 1668, 1642 cm⁻¹, 1H-NMR (CDCl₃) 1 93 (s, 3H), 4 51 (d, J = 6 7 Hz, 2H), 4 74 (d, J = 6 7 Hz, 2H), 6 22 (dt, J = 15 9 and 6 7 Hz, 1H), 6 32 (dt, J = 15 9 and 6 7 Hz, 1H), 6 62 (d, J = 15 9 Hz, 1H), 6 72 (d, J = 15 9 Hz, 1H), 7 02 (s, 1H), 7 20-7 30 (m, 10H), 13C-NMR (CDCl₃) 12 9, 42 9, 50 3, 109 9, 122 6, 122 8, 126 3, 126 4, 127 5, 128 1, 128 3, 128 5, 134 4, 135 5, 136 4, 137 6, 151 0, 163 2 <u>Anal</u> Calcd for C₂₃H₂₂N₂O₂ C, 77 07, H, 6 19, N, 781 Found C, 76 17, H, 6 17, N, 7 53

19 m p 168-9C (from diethyl ether), IR(KBr) 1713, 1641 cm⁻¹, 1H-NMR (CDCl₃) 1 95 (s, 3H), 4 69 (d, J = 6 7 Hz, 2H), 6 27 (dt, J = 15 8 and 6 7 Hz, 1H), 6 66 (d, J = 15 8 Hz, 1H), 7 04 (d, J = 6 1 Hz, 1H), 7 18-7 39 (m, 5H), 10 42 (N-H, 1H), 13C-NMR (CDCl₃) 13 8, 42 3, 110 1, 122 7, 126 4, 127 7, 128 4, 133 8, 134 7, 136 4, 153 1, 163 7 <u>Anal</u> Calcd for C₁₄H₁₄N₂O₂ C, 69 40, H, 5 82, N, 11 56 Found C, 68 92, H, 5 86, N, 11 43

1-Cinnamyluracil, 21, 3-cinnamyluracil, 22, and 1,3-dicinnamyluracil, 23 (Run 12) This experiment was performed as for compounds 18-20. Compounds 21-23 were separated by the same procedure.

21 m p 199-200C, IR(KBr) 1714, 1672 cm⁻¹, 1H-NMR (d6-DMSO) 4.43 (d, J = 61 Hz, 2H), 5 59 (d, J = 79 Hz, 1H), 6 32 (dt, J = 159 and 6 1 Hz, 1H), 6 55 (d, J = 159 Hz, 1H), 7 24-7 43 (m, 5H), 7 64 (d, J = 79 Hz, 1H), 11 28 (s, 1H), 13C-NMR (d6-DMSO) 48 9, 101 4, 124 4, 126 6, 128 0, 128 8, 132 6, 136 1, 145 4, 151 0, 163 9; MS(m/e) 228(M, 34), 117(100), 115(49), 91(18) Anal Calcd for C13H12N2O2 C, 68 41, H, 5 30, N, 12 27 Found C, 68 43, H, 5 31, N, 12 20

23 m p 101-2C, IR(KBr) 1701, 1668, 1647 cm⁻¹, 1H-NMR (d₆-DMSO) 4 52 (d, J = 61 Hz, 2H), 4 57 (d, J = 49 Hz, 2H), 5 77 (d, J = 73 Hz, 1H), 6 26 (dt, J = 159 and 6 1 Hz, 1H), 6 35 (d, J = 159 and 6 1 Hz, 1H), 6 49 (d, J = 159 Hz, 1H), 6 58 (d, J = 159 Hz, 1H), 7 19-7 44 (m, 10H), 7 73 (d, J = 73 Hz, 1H), 13C-NMR (d₆-DMSO) 42 2, 50 2, 100 7, 124 1, 124 2, 126 4, 126 6, 127 8, 128 1, 128 7, 128 8, 132 0, 132 8, 136 1, 136 3, 144 1, 150 9, 162 4 <u>Anal</u> Calcd for C₂₂H₂₀N₂O₂ C, 76 72, H, 5 85, N, 8 13 Found C, 76 80, H, 5 86, N, 8 16

22 m p 160-1C, IR(KBr) 1739, 1634, 1608 cm⁻¹, 1H-NMR (d6-DMSO) 4 50 (d, J = 6 1 Hz, 2H), 5 61 (d, J = 8 6 Hz, 1H), 6 24 (dt, J = 15 8 and 6 1 Hz, 1H), 6 47 (d, J = 15 8 Hz, 1H), 7 19-7 50 (m, 6H), 11 14 (s, 1H), 13C-NMR (d6-DMSO) 41 2, 103 9, 124.3, 126 4, 127 8, 128 8, 131 8, 136 4, 140 9, 151 4, 163 0 <u>Anal</u> Calcd for C13H12N2O2 C, 68 41, H, 5 30 Found C, 68 10, H, 5 36

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